Temperature Dependence of Gating Current in Myelinated Nerve Fibers

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Summary. Asymmetrical displacement currents and Na currents of single myelinated nerve fibers of Xenopus laevis were studied in the temperature range from 5 to 24°C. The time constant of the on-response at E = 4 mV, τ_{on} , was strongly temperature dependent, whereas the amount of displaced charge at E = 39 mV, Qon, was only slightly temperature dependent. The mean Q10 for τ_{on}^{-1} was 2.54, the mean Q_{10} for Q_{on} was 1.07. The time constant of charge immobilization, τ_i , at E = 4 mV varied significantly (α = 0.001) with temperature. The mean Q_{10} for τ_i^{-1} was 2.71 \pm 0.38. The time constants of immobilization of gating charge and of fast inactivation of Na permeability were similar in the temperature range from 6 to 22°C. The Q_{off}/Q_{on} ratio for E = 4 mV pulses of 0.5 msec duration decreased with increasing temperature. The temperature dependence of the time constant of the off-response could not be described by a single Q_{10} value, since the Q_{10} depended on the duration of the test pulse. Increasing temperature shifted $Q_{on}(E)$ curves to more negative potentials by 0.51 mV K^{-1} , but shifted $P_{\text{Na}}(E)$ curves and $h_{\infty}(E)$ curves to more positive potentials by 0.43 and 0.57 mV K⁻¹, respectively, h_{∞} (E = -70 mV) increased monotonously with increasing temperature. The present data indicate that considerable entropy changes may occur when the Na channel molecule passes from closed through open to inactivated states.

Key Words myelinated nerve fiber · gating current · temperature

Introduction

Since the first measurement on nerve membrane by Armstrong and Bezanilla (1973), asymmetry currents have been demonstrated to be, at least mostly, associated with the opening and closing of Na channels. Therefore, it seems to be justified to use the terms "asymmetry current" and "gating current" synonymously. The relation between activation of macroscopic current and charge movement, on one hand, and between inactivation of macroscopic current and charge immobilization, on the other hand, may be rather complex. Gating current reflects mainly reaction steps before, but also after the one that opens the Na channel (Neumcke, Nonner & Stämpfli, 1976). The role of charge immobilization remains unclear. Armstrong and Bezanilla (1977) reported that kinetics and potential dependence of immobilization and inactivation are similar, whereas Meves and Vogel (1977) found that immobilization develops four times slower than inactivation. Nonner (1979) reported that both processes are similarly affected by a scorpion venom, whereas Tanguy and Yeh (1988) observed that batrachotoxin removes inactivation but enhances charge immobilization. Thus, it is most likely that charge immobilization and inactivation are linked, but reflect different molecular steps (Nonner, 1980).

The effect of temperature on gating current was first investigated by Bezanilla and Taylor (1978) and Kimura and Meves (1979) on squid axon. The temperature dependence of charge immobilization has not been characterized yet. Whereas it is accepted that steady-state parameters of Na permeability are shifted when temperature is changed (Dudel & Rüdel, 1970; Kimura & Meves, 1979; Chiu, Mrose & Ritchie, 1979; Schwarz, 1986), little has been known about how stationary gating current parameters change with temperature. Altogether, only two short reports deal with the influence of temperature on gating current in amphibian nerve fibers (Nonner, Rojas & Stämpfli, 1975; Collins & Rojas, 1982). The aim of the present study was to measure temperature dependence of gating current and Na current of the node of Ranvier under similar conditions and to integrate these data into a thermodynamic model of Na channel gating. Previous findings on squid axon are extended in regard to temperature dependence of charge immobilization and steadystate parameters. It is concluded that considerable entropy changes may be associated with conformational changes of the Na channel protein. A preliminary account of some of the experiments has been published elsewhere (Jonas & Vogel, 1988).

Materials and Methods

Single myelinated nerve fibers were dissected from the motor or sensory branch of the tibial nerve of decapitated clawed toads (Xenopus laevis). The fibers were mounted in a Perspex chamber, fixed with Vaseline or Glisseal (Borer Chemie, Solothurn, Switzerland) seals, and held under current or voltage-clamp conditions (Nonner, 1969). Both ends of the fiber were cut as short as possible in order to increase the bandwidth of the recordings (Nonner, Rojas & Stämpfli, 1978). The absolute membrane potential, where h_{x} was 0.6–0.8 at about 12°C, was assumed to be E = -70 mV and the holding potential was adjusted to E = -100mV throughout the experiment. The preparation was allowed to stabilize before the measurement began. Membrane current values, I_m , were calculated from the voltages at the compartment E as $I_m = V_E/R_{ED}$, where R_{ED} , the axoplasmic resistance between compartment E and the node, was estimated from the actual fiber geometry and a specific axoplasmic resistance of 100 Ω cm. The decrease of R_{ED} with increasing temperature with a Q₁₀ of 1.22 (Neumcke & Stämpfli, cited in Schwarz, 1986) or about 1.3 (Chiu et al., 1979) was not taken into account. The current signals were filtered by a 4-pole low-pass Bessel filter with corner frequencies of 20 or 28 kHz (3 dB attenuation).

The temperature of the preparation was measured with a thermocouple in the nodal compartment A less than 2 mm distant from the node and was continuously recorded with a multi-pen recorder. The accuracy of this kind of temperature measurement was estimated to be about $\pm 0.3^{\circ}$ C. Two Peltier elements cooled or heated the aluminium block surrounding the nerve chamber and the solution flowing through the nodal compartment. The temperature was changed slowly (by about 1°C min⁻¹) to avoid hysteresis effects (*see* Schwarz, 1979).

A Sirius 1 microcomputer (Sirius Systems Technology, Scotts Valley, CA 95066) generated pulses via a 12-bit digital-toanalog converter and sampled membrane currents via a 12-bit analog-to-digital converter at a rate of a multiple of 5 μ sec (de Haas, 1987). Gating currents and Na currents were elicited with the same pulse program. It consisted of a 50-msec prepulse with E = -120 mV, a test pulse with $E = E_{\text{test}}$ of 0.25 to 30 msec duration and a 2-msec postpulse with E = -120 mV. For measurement of gating currents, most of the leakage and capacity current was subtracted analogically and the residual symmetrical current was subtracted digitally by means of a +P/2 procedure, i.e., with two voltage jumps from $E = (-120 \text{ mV} - \Delta E/2)$ to E =-120 mV, where $\Delta E = E_{\text{test}} - (-120 \text{ mV})$. For measurement of Na currents, leakage and capacity currents were estimated from a negative pulse and subtracted digitally. Usually, 1 to 16 (Na currents) and 32 and 448 (asymmetry currents) single sweeps were averaged to improve the signal-to-noise ratio. The interpulse interval was chosen long enough (0.9 sec) to allow sufficient recovery from inactivation. Charge us. voltage curves, Qon (E), were obtained from two runs, one with increasing pulse potential and the other one with decreasing pulse potential; corresponding records were finally averaged. This procedure was chosen to avoid systematical errors, which could arise if, for instance, Qon values at lower potentials are always measured before the Qon values at higher potentials. Similarly, immobilization curves were obtained from one run with increasing pulse length and another run with decreasing pulse length. The error of charge due to the recording apparatus, Q_{app}, was determined with a realistic dummy circuit representing the nerve fiber and the seal resistances (membrane capacity $C_m = 8.2$ pF, series resistance $R_s = 1.0 \text{ M}\Omega$, $R_{ED} = 3.3 \text{ M}\Omega$). Q_{app} was estimated to be 4.2 fC for the on-charge moving during a 0.5-msec pulse with E = 4 mV. Compared to the gating current of the Na channels (see results), this is about two orders of magnitude smaller.

In order to take the delay of the current signal introduced by the low-pass filter into account, the time axis of the current recordings was shifted by 15 μ sec with respect to that of the pulses (Nonner et al., 1978). Time constants τ of the on- and offresponses of the asymmetry current were obtained after integration and fit with the function

$$Q(t) = Q_{\alpha} \cdot [1 - \exp(-t/\tau)] + I \cdot t.$$
⁽¹⁾

The linear term $I \cdot t$ allows for very slow components of charge movement (e.g., off-response components due to the return of immobilized charge) or components of constant current (e.g., the "pedestal" of on-responses). Usually, only data points with 0.02 msec $\leq t \leq 1$ msec for the on-responses and 0.02 msec $\leq t \leq 0.3$ msec for the off-responses were taken into account. The amount of displaced charge was either obtained from fit with Eq. (1) or by integration for $0 \leq t \leq 0.5$ msec. The last method required the subtraction of the pedestal, which was measured between 0.5 and 1 msec after the onset of the pulse. Displaced charge is always given as a positive value regardless of the direction of the current flow.

Time constants of activation and inactivation of Na permeability were obtained from fit with the function

$$I_{\text{Na}}(t) = \begin{cases} \{1 - \exp[-(t - \delta t)/\tau_m]\}^2 \cdot [A \cdot \exp(-t/\tau_h) \\ + B] \text{ for } t \ge \delta t \\ 0 \text{ for } t < \delta t \end{cases}$$
(2)

(see Neumcke et al., 1976). Usually, only data points with 0.02 msec $\leq t \leq 5$ msec were taken into account. Instead of a slow exponential component of inactivation (Kniffki, Siemen & Vogel, 1981), a time-independent component *B* was introduced. This simplification accelerated the convergence of the fit algorithm remarkably. The τ_h values obtained with Eq. (2) are nearly equal to the τ_{h1} values of Kniffki et al. (1981; *compare* their Fig. 3 with Fig. 4*B* of the present paper).

 $Q_{on}(E)$ curves, $P_{Na}(E)$ curves, and $h_{\alpha}(E)$ curves (see Fig. 6) were fitted with the function

$$Y(E) = Y_{\text{max}} / \{1 + \exp[(E_h - E)/k^*]\}, \ Y = Q_{\text{on}}, \ P_{\text{Na}}, \ \text{or} \ h_{\infty}, \quad (3)$$

where E_{h} is the midpoint potential of the curve and k^{*} is the slope factor. Y_{max} was set to 1 for $Y = h_{\infty}$.

Nonlinear least squares fits were performed by means of an improved Gauss-Newton method (Jennrich & Ralston, 1979), which also provided the standard error of each estimated parameter. Mean values (e.g., in Tables 1 and 2) are given as mean \pm SEM unless otherwise noted. In some plots (Figs. 4 and 7), points within a \pm 0.5°C temperature interval are condensed to one mean value in order to increase the clarity of the illustrations. Error bars indicate the SEM if it exceeds the size of the symbol. The statistical significance of a correlation between two parameters was estimated from the value of the correlation coefficient according to Sachs (1984).

During the measurement of asymmetry currents, the node was continuously superfused with a standard solution containing (in mM) 107 tris(hydroxymethyl)-aminomethane (Tris) chloride, 2.0 CaCl₂, 10 tetraethylammoniumchloride (TEA), and 500 nM tetrodotoxin (TTX), pH = 7.4 ± 0.05 . In some experiments, 97 mM tetramethylammoniumchloride and 10 mM TrisCl were used instead of 107 mM TrisCl. During the measurement of Na currents, the fiber was superfused with a Ri-TEA solution contain-

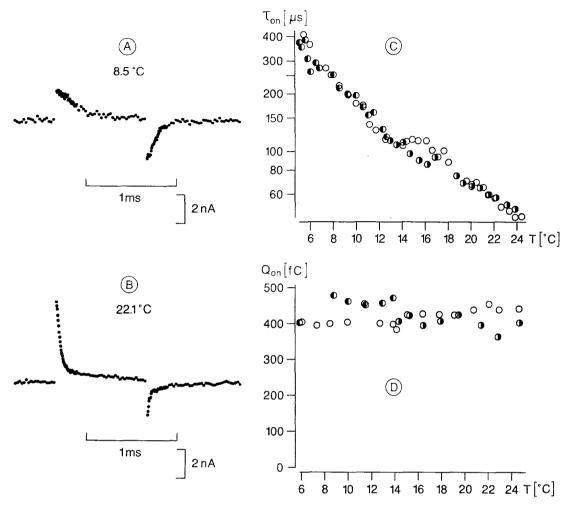


Fig. 1. Temperature dependence of asymmetry current kinetics and amount of displaced charge. (A, B) Original recordings are shown of asymmetry currents at 8.5°C (A) and 22.1°C (B). Current responses to 32 pulses were averaged. The first five sample points after each voltage step, corresponding to 25 μ sec, are not plotted. Corner frequency of the low-pass filter was 28 kHz; test pulse E = 4 mV. (C) Time constant of the on-response of the asymmetry current, τ_{on} , was plotted logarithmically vs. temperature. τ_{on} was obtained from fit with Eq. (1). Test pulse E = 4 mV. (D) Amount of displaced charge, Q_{on} , was plotted vs. temperature. Q_{on} was obtained from direct integration of the first 0.5 msec of the on-response after subtraction of the pedestal (0.5–1 msec). Test pulse E = 39 mV. Data in C and D are from one temperature cycle each; the sequence of measurements was: from intermediate to low (\mathbb{O}), from low to high (\bigcirc), and from high to intermediate temperature (\mathbb{O}). Expt. TE805

ing (in mM) 110 NaCl, 2.5 KCl, 2.0 CaCl₂, 5.0 N,N-bis(2-hydroxvethyl)-2-aminoethane sulfonic acid (BES), 10 TEA, pH adjusted to 7.4 \pm 0.05 with Tris base. TTX (5 nm) was added in most of the Na current experiments to minimize artifacts due to series resistance, R_s . R_s could pretend shifts of $P_{Na}(E)$ curves, because the peak current increases with temperature. The precise estimation of the amount of shift due to R_s is, however, difficult, because R_s may be temperature dependent itself. Assuming the most unfavorable case of a temperature-independent $R_s = 280 \text{ k}\Omega$ (Drouin & Neumcke, 1974), $P_{\text{Na}} = 3 \times 10^{-9} \text{ cm}^3$ sec⁻¹ (25°C), $P_{\text{Na}} = 1.5 \times 10^{-9} \text{ cm}^3 \text{ sec}^{-1}$ (5°C), a reversal potential $E_{\text{Na}} = 60 \text{ mV}$, and $E_h = -40 \text{ mV}$ (see Fig. 6B), one can calculate a P_{Na} shift of 4 mV to the left when temperature is increased. The cut ends of the fiber were bathed in a solution containing, in mM, 110 CsF, 10 TrisCl, pH = 7.2 ± 0.05 . Control measurements revealed that electrode potentials and junction potentials are constant within about $\pm 1 \text{ mV}$ when temperature is changed from 5 to 25°C.

Results

TEMPERATURE DEPENDENCE OF au_{on} and Q_{on}

Figure 1A and B show asymmetry currents recorded at 8.5°C and 22.1°C. The test pulse potential was E = 4 mV, the potential before and after the testpulse was E = -120 mV. The kinetics of onresponse and off-response of the asymmetry current are both faster at 22.1°C as compared with 8.5°C. Although the maximum asymmetry current is higher at higher temperature, the amount of charge Q_{on} carried during the first 0.5 msec of the on-response after subtraction of the pedestal is similar in both cases (Q_{on} = 414 fC at 8.5°C, Q_{on} = 423 fC at 22.1°C).

Expt.	Q ₁₀	Temperature range (°C)	Significant correlation ^a
AS724	1.06	5.6-21.2	yes
AS726	1.11	6.2-19.1	no
AS727	1.27	4.3-21.1	yes
AS730	0.92	8.2-18.7	no
TE805	0.99	5.9-24.7	no
Mean ± SEM	$\begin{array}{c} 1.07 \\ \pm 0.06 \end{array}$		

Table 1. Temperature dependence of Qon

^a Two-sided test, $\alpha = 0.05$.

The effect of temperature on the kinetics of the on-response is quantitatively analyzed in Fig. 1C. The time constant of the on-response, τ_{on} , was obtained by fitting integrated on-responses with Eq. (1) and was plotted logarithmically *vs.* temperature. For $t \leq 1$ msec, the time course of the on-responses could be sufficiently described by one exponential function and a linear term (correlation coefficient, *r*, between 0.9999 and 0.9913). At temperatures above 20°C, a second slower component was sometimes visible, but was not analyzed further. The mean Q₁₀ value for τ_{on}^{-1} obtained from eight experiments was 2.54 ± 0.07 (*see* legend of Fig. 7) in the temperature range 5–24°C.

In Fig. 1D, the amount of charge Q_{on} carried during the first 0.5 msec of a test pulse with E = 39mV was plotted vs. temperature. The high testpulse potential was chosen to give an estimate of the maximum amount of displaced charge. Qualitatively, Q_{on} does not increase with temperature; it even decreases slightly in this experiment. Table 1 lists the Q_{10} values for Q_{on} obtained from five experiments. A significant variation of Q_{on} with temperature was only seen in two of these experiments. The mean Q_{10} value for Q_{on} at E = 39 mV is 1.07 ± 0.06 .

TEMPERATURE DEPENDENCE OF τ_m

Frankenhaeuser and Moore (1963) reported surprisingly low Q_{10} values for the activation kinetics of Na permeability. It therefore seemed interesting to reinvestigate this point with improved techniques of data acquisition and evaluation. Figure 2A shows recordings of Na currents at 8.3 and 21.9°C. TTX (5 nM) was added to the extracellular solution in order to reduce series resistance artifacts. The Na current recordings of Fig. 2A may be compared with the gating current recordings of Fig. 1A and B, because prepulse potential (-120 mV), testpulse potential (4 mV), internal solution, and temperature are similar. The kinetics of activation and inactivation is faster

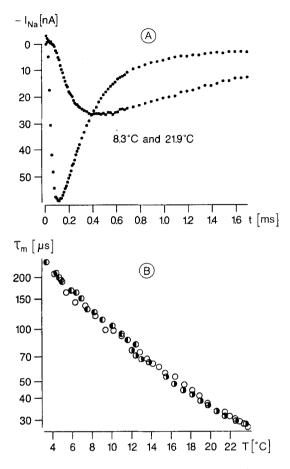


Fig. 2. Temperature dependence of Na current kinetics. (A) Na currents are shown at 8.3 and 21.9°C in Ri-TEA + 5 μ M TTX. The registration with the larger peak current was recorded at higher temperature. Current responses to 16 pulses were averaged. Corner frequency of the low-pass filter was 28 kHz (Expt. TE802). (B) Time constant of Na permeability activation, τ_m , is shown plotted logarithmically *vs.* temperature. τ_m was obtained from fit with Eq. (2). Data are from one temperature cycle; the different symbols have the same meaning as in Fig. 1 (Expt. TE804). Test pulse E = 4 mV throughout the figure

and the peak current is larger at 21.9 than at 8.3°C. The time constant, τ_m , of the Na permeability activation was plotted logarithmically *vs*. temperature in Fig. 2*B*. τ_m values were obtained by fitting Na current recordings with Eq. (2), which includes an exponent of two and an additional delay term (Neumcke et al., 1976). The mean Q₁₀ value for τ_m^{-1} obtained from five experiments was 2.64 ± 0.06 (see legend of Fig. 7) in the temperature range 5–24°C.

The present data indicate that the Q_{10} values for τ_{on}^{-1} of the gating current and the Q_{10} values for τ_m^{-1} of the Na current are not significantly different. Figures 1*C* and 2*B* present data obtained from one cyclic change of temperature each. Since the τ value at the beginning of the temperature cycle is equal to

the τ value at the end of the temperature cycle, the effects of temperature on τ_{on} and τ_m seem to be fully reversible. The Q₁₀ value for τ_{on}^{-1} and τ_m^{-1} is slightly larger at low temperature than at high temperature. A slight temperature dependence of the Q₁₀ value is, however, expected even if activation enthalpy is constant (*see* Kniffki et al., 1981). There is no evidence for a phase transition in the temperature range 5–24°C (Schwarz, 1979).

TEMPERATURE DEPENDENCE OF CHARGE IMMOBILIZATION

Figure 3 shows the temperature dependence of charge immobilization kinetics. Q_{off}/Q_{on} is plotted vs. the length of the immobilizing pulse (E = 4 mV) at 5.8 and 6.2°C in Fig. 3A and at 18.9°C in Fig. 3B. Charge immobilization develops faster at higher temperature. The points can be satisfactorily fitted with one exponential function and a component of nonimmobilizable charge Q_n (see legend of Fig. 3). $Q_n/(Q_i + Q_n)$ was slightly higher at higher temperature, but its temperature dependence was not significant (r = 0.52, 12 degrees of freedom, two-sided test, $\alpha = 0.05$). $Q_n/(Q_i + Q_n)$ estimated from 14 immobilization curves at various temperatures is $21.6 \pm 8.25\%$ (mean value \pm sp, eight fibers). If charge immobilization has a temperature-dependent kinetics, one would expect that the Q_{off}/Q_{op} ratio for pulses of constant length varies with temperature. Figure 3C shows that this is the case. The Q_{off}/Q_{on} ratio for 0.5-msec pulses decreases monotonously from values of about 0.8 at 6°C to values of approximately 0.4 at 22°C.

Figure 4A shows the time constant of charge immobilization, τ_i , plotted logarithmically vs. temperature. τ_i values were obtained with two different methods: (i) $Q_{off}/Q_{on}(t_{pulse})$ was fitted with an exponential function as shown in Fig. 3A and B. (ii) Alternatively, τ_i was obtained from the Q_{off}/Q_{on} ratio for a pulse of constant length (see Fig. 3C): If $Q_{off}/Q_{on}(t_{pulse}) = Q_i \cdot \exp(-t_{pulse}/\tau_i) + Q_n$ and if one assumes that Q_i and Q_n do not change with temperature, it follows that

$$\tau_i = -t_{\text{pulse}}/\ln[(Q_{\text{off}}/Q_{\text{on}} - Q_n)/Q_i].$$
(4)

Hence, τ_i can be directly obtained from the Q_{off}/Q_{on} ratio if Q_i and Q_n are determined once in each experiment. Method *i* allows a more precise estimation of τ_i , but it is not possible to measure τ_i at more than two or three temperatures in one fiber. Method *ii* makes it possible to determine τ_i for a lot of temperatures in one fiber, but produces considerable scatter of data, because each τ_i depends on only one Q_{off}/Q_{on} value. It may lead to incorrect results if Q_i

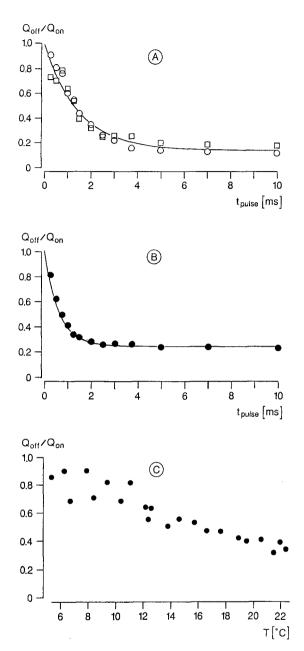
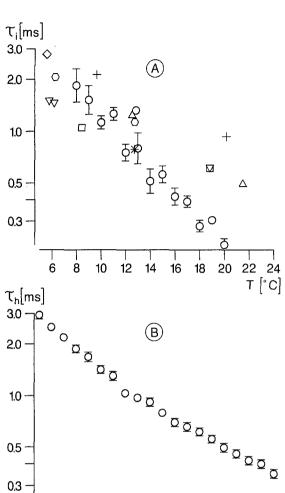


Fig. 3. Temperature dependence of charge immobilization kinetics. (A) Relative off-response charge, Q_{off}/Q_{on} , is shown plotted vs. pulse length, t_{pulse} , at 6.2°C (O) and 5.8°C (D), (B) at 18.9°C (•). $Q_{off}(t_{pulse})$ was obtained by fitting the off-response (0.3 msec) after pulses of duration t_{pulse} with Eq. (1). $Q_{on}(t_{pulse})$ was obtained by averaging all corresponding on-responses and fitting them with Eq. (1). The continuous curves are drawn according to least squares fits with the function $Q_{off}/Q_{on}(t_{pulse}) = Q_i \cdot \exp(-t_{pulse}/\tau_i)$ + Q_n to the mean of the points (O) and (D) in A and the points (•) in B. $\tau_i = 1.47$ msec and $Q_n = 0.138$ in A, $\tau_i = 0.61$ msec and $Q_n = 0.246$ in B. Points at $t_{pulse} = 15$ msec were included into the fit, but are not displayed in the figure. Sequence of the measurements was (A \bigcirc), (B \bullet), (A \square) (Expt. AS726). (C) Q_{off}/Q_{on} was obtained with test pulses of a constant duration of 0.5 msec. Q_{off} was obtained as in A and B, but Q_{on} was obtained by direct integration of the on-response, because the pulses are short (Expt. AS740). Immobilizing pulse E = 4 mV throughout the figure



24

22 24

T [°C]

Fig. 4. Temperature dependence of immobilization time constant compared with fast inactivation time constant. (A) Time constants of charge immobilization, τ_i , are shown plotted logarithmically vs. temperature. Circles: τ_i was obtained from Q_{off}/ Qon according to Eq. (4), data from 81 gating current records (two experiments), test-pulse length 0.25, 0.5, 0.75, and 1.0 msec. All other symbols: τ_i was obtained from fit of Q_{off}/Q_{on} (t_{pulse}) as in Fig. 3A and B, data from eight experiments; different experiments are represented by different symbols. The symbol (∇) corresponds to the experiment of Fig. 3A, B. Immobilizing pulse E = 4 mV. (B) Time constants of fast inactivation of Na permeability, τ_h , are shown plotted logarithmically vs. temperature. Data are from 394 Na current records (five experiments). Test pulse E = 4 mV

and Q_n are not temperature independent and if Q_{off} $Q_{on}(t_{pulse})$ is not sufficiently described by a single exponential term. A slight temperature dependence of Q_n indeed cannot be excluded (see above). In Fig. 4A, the circles represent mean τ_i values obtained with method *ii*, all other symbols represent data obtained with method i. The mean Q_{10} value for τ_i^{-1} is 2.71 \pm 0.38 if the τ_i values obtained with method *i* are weighted with a factor of 2 because of

their higher reliability. The correlation between τ_i and temperature is highly significant (r = -0.81, 25degrees of freedom, two-sided test, $\alpha = 0.001$). For comparison, Fig. 4B shows the time constant of Na permeability inactivation, τ_h , plotted logarithmically vs. temperature. τ_h was obtained from fit of Na currents with Eq. (2). The mean Q_{10} value for τ_h^{-1} is 3.00 ± 0.09 (see legend of Fig. 7). Thus, the Q₁₀ values for τ_i^{-1} and τ_h^{-1} are not significantly different. Moreover, τ_i and τ_h are similar within the investigated temperature range 6-22°C.

Temperature Dependence of $au_{ m off}$

Figure 5A and B shows integrated off-responses after a voltage jump from E = 4 mV to E = -120 mVat 6.2 and 18.9°C. The duration of the test pulses was 0.25, 0.75, 1.5 and 2.5 msec. The kinetics of the off-response is faster at higher temperature. Figure 5C displays the time constants of the off-response at 6.2 and 18.9°C obtained from fit with Eq. (1). At low temperature, $\tau_{\rm off}$ increases with increasing testpulse duration, reaches a maximum value at 0.75 msec, and then decreases again. At high temperature, the rising phase is less pronounced. A similar dependence of τ_{off} on pulse length was reported by Dubois and Schneider (1982). The authors suggested that τ_{off} increases with increasing amount of Qon and decreases as charge immobilization develops. Figure 5C shows clearly that the determination of the Q₁₀ value for τ_{off}^{-1} with pulses of a constant duration may give different values, depending on the length of the pulse. 0.25-msec pulses yield a Q_{10} of only 1.74, whereas 1.5-msec pulses give a Q_{10} of 2.52. Such a dependence of Q_{10} on pulse duration is in accordance with the suggestions of Dubois and Schneider (1982), since the maximum of $\tau_{\text{off}}(t_{\text{pulse}})$ would be shifted towards lower t_{pulse} values when on-response kinetics and immobilization kinetics become faster as temperature is increased.

TEMPERATURE DEPENDENCE OF $Q_{on}(E)$, $P_{Na}(E)$, and $h_{\infty}(E)$

Figure 6A-C shows the effect of temperature on stationary gating parameters as functions of potential. The filled circles represent measurements at about 25°C; the open circles symbolize measurements at approximately 5°C. The half-filled symbols represent the average from three measurements at about 13°C, which have been performed before, between, and after the measurements at 5 and 25°C.

In Fig. 6A, the on-response charge, Qon, was plotted vs. test-pulse potential E for different temperatures. $Q_{on}(E)$ was fitted with Eq. (3) and all Q_{on} values were normalized to the maximum charge $(Y_{\text{max}} \text{ in Eq. (3)})$. Increasing temperature by 18.6°C shifts $Q_{on}(E)$ curves towards negative potentials by

0.3

6 8 10 12 14 16 18 20

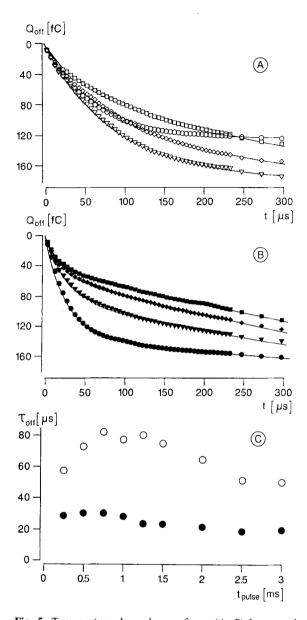


Fig. 5. Temperature dependence of τ_{off} . (A, B) Integrated offresponses of asymmetry currents are shown at 6.2°C (A) and 18.9°C (B). Length of the immobilizing pulse: $t_{\text{pulse}} = 0.25$ msec (circles), 0.75 msec (triangles), 1.5 msec (diamonds), and 2.5 msec (squares). Zero-current line was estimated from 39 samples (1 msec) directly before the onset of the test pulse and 39 samples (1 msec) 1 msec after the end of the test pulse. The continuous curves were obtained from fit with Eq. (1). (C) Time constant of the integrated off-responses, τ_{off} , was plotted *vs*. length of the immobilizing pulse, t_{pulse} , at 6.2°C (O) and 18.9°C (\bullet). Potential of the immobilizing pulse E = 4 mV, potential after the pulse E = -120 mV. (Expt. AS726)

11.8 mV. The estimation of the midpoint potentials seems to be quite reliable, because the $Q_{on}(E)$ curves at high as well as at low temperature saturate in the positive potential range. In Fig. 6B, Na permeability, P_{Na} , is plotted logarithmically vs. testpulse potential. Increasing temperature by 20.0°C shifts $P_{Na}(E)$ towards positive potentials by 8.3 mV.

Table 2.

A. Temperature dependence of midpoint potential of $Q_{cn}(E)$ curves

Expt.	Shift (mV)	Temperature range (°C)	Slope factor $ k^* $ (mV) low/high temp.
AS719	-8.3	5.6-24.8	15.2/22.5
AS724	-8.6	5.6-21.1	13.1/15.1
AS726	-9.2	6.0-19.1	18.0/22.4
AS734	-2.6	7.4-19.3	24.0/20.5
TE805	-11.8	6.1-24.7	15.3/18.0
Mean	_	$-0.508 \text{ mV } K^{-1}$	18.41
\pm sem	±	0.085	± 1.19

B. Temperature dependence of midpoint potential of $P_{\text{Na}}(E)$ curves^a

Expt.	Shift (mV)	Temperature range (°C)	Slope factor $ k^* $ (mV) low/high temp.
TE804	5.2	3.9–24.6	5.5/4.6
TE806	8.3	4.3-24.3	7.2/5.3
TE810	10.2	5.6-21.9	4.1/3.9
Mean		0.429 mV K ⁻¹	5.10
\pm sem	<u>+</u>	0.109	± 0.49

C. Temperature dependence of midpoint potential of $h_{\infty}(E)$ curves

Expt.	Shift (mV)	Temperature range (°C)	Slope factor $ k^* $ (mV) low/high temp.
TE804	13.9	4.1-25.4	7.3/5.5
TE806	10.4	4.2-24.3	8.7/7.2
TE807	7.2	5.2-23.3	8.1/5.8
TE810	11.9	5.3-22.0	9.5/7.8
Mean		$0.570 \text{ mV } K^{-1}$	7.49
\pm sem	<u>+</u>	0.070	± 0.48

^a Midpoint potentials of the $P_{Na}(E)$ curves were obtained from semilogarithmic plots as in Fig. 6B; a logarithmic form of Eq. (3) was used for the fit.

Figure 6C shows steady-state inactivation, h_{∞} , plotted *vs*. prepulse potential. The length of the prepulses was 50 msec for the curve at 25.4°C and 250 msec for the curve at 4.1°C to compensate for the slower inactivation kinetics at lower temperature. The $h_{\infty}(E)$ curve is shifted towards positive potentials by 13.9 mV when temperature is increased by 21.3°C.

As the 95% confidence intervals of the midpoint potentials do not overlap (see legend of Fig. 6), the difference between E_h at low and high temperature is significant for all curves. The shifts seem to occur in the whole temperature range 5–25°C, because the points at intermediate temperature are between the curves at low and high temperature. Shifts of $Q_{on}(E)$, $P_{Na}(E)$, and $h_{x}(E)$ observed in nine experiments including the data of Fig. 6 are summarized in Table 2.

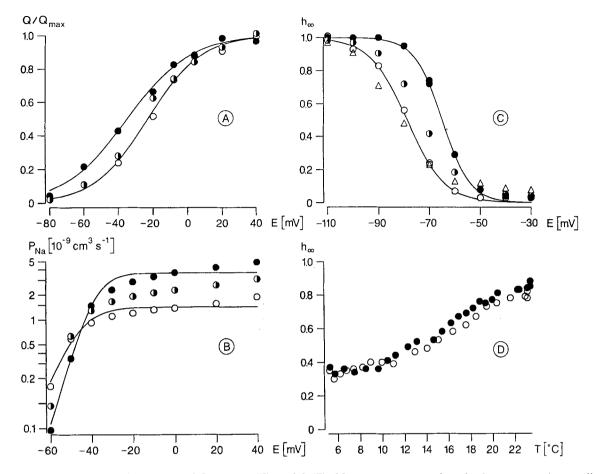


Fig. 6. Temperature dependence of $Q_{on}(E)$, $P_{Na}(E)$, and $h_x(E)$. Measurements were done in the sequence intermediate—low intermediate—high—intermediate temperature. Data from the three measurements at intermediate temperature were averaged and displayed as one symbol (**①**) for each potential value. Na currents were recorded in the presence of 5 nM TTX. (A) $Q_{on}(E)$ curves at 24.7°C (**①**), intermediate temperature (14.2, 13.9, and 13.3°C, **①**), and 6.1°C (**○**). Q_{on} was obtained from fit of the integrated onresponses with Eq. (1), and $Q_{on}(E)$ was fitted with Eq. (3). All Q_{on} values were normalized to the maximum charge (Y_{max} in Eq. 3) (Expt. TE805). (B) $P_{Na}(E)$ curves at 24.3°C (**①**), intermediate temperature (12.8, 13.2, and 13.8°C, **①**), and 4.3°C (**○**). The P_{Na} values were obtained from the peak current values according to the Goldman-Hodgkin-Katz equation (Expt. TE806). (C) $h_x(E)$ curves at 25.4°C (**①**), intermediate temperature (13.0, 13.3, and 13.0°C, **①**), and 4.1°C (**○**, Δ). The h_x values at low temperature were obtained with 250msec (**○**) and 50-msec (Δ) prepulses. All other h_x values were obtained with 50-msec prepulses. Test pulse E = 0 (Expt. TE804). Continuous curves are drawn according to least squares fits of the data at low temperature (**○**) and high temperature (**●**), respectively, with Eq. (3) and correspond to the following parameters:

	$Q_{on}(E)$ curve low/high temp.	$P_{\text{Na}}(E)$ curve low/high temp.	$h_{x}(E)$ curve low/high temp.
$E_h(mV)$	-22.5/-34.3	-45.6 /-37.3	-78.5/-64.6
	$\pm 1.8 \pm 1.8$	\pm 2.3 \pm 1.8	\pm 0.3 \pm 0.3
$ k^* $ (mV)	15.3/ 18.0	7.2 / 5.3	7.3/ 5.5
Ymax	453 / 381 fC	1.43/ 3.67	1
		imes 10 ⁻⁹ cm ³ sec ⁻¹	

(D) $h_{x}(E = -70 \text{ mV})$ is plotted vs. temperature. Symbols (\bigcirc) represent measurements during decrease of temperature, symbols (\bigcirc) represent h_{x} values obtained while increasing temperature. 50-msec prepulse, test pulse E = 0. (Expt. TE807)

In Figure 6D, $h_{\infty}(E = -70 \text{ mV})$ is plotted vs. temperature for one temperature cycle. h_{∞} increases reversibly with increasing temperature in the range from 5 to 23°C. If the midpoint potential, E_h , varies linearly with temperature, $h_{\infty}(E = \text{const} = -70 \text{ mV})$, T) would be expected to show the same sigmoidal shape and should be described by the same mathematical function as an $h_{\infty}(E, T = \text{const})$ curve, namely $h_{\infty}(E = -70 \text{ mV}, T) = 1/\{1 + \exp[(T - T_h)/k^{**}]\}$. T_h is the temperature where h_{∞} is 0.5 and k^{**}

is a slope factor, which characterizes the steepness of the temperature dependence of E_h . The general shape of $h_{\infty}(E = -70 \text{ mV}, T)$ is indeed sigmoidal, but h_{∞} saturates at values >0 for $T \le 6^{\circ}$ C. This deviation may be due to the fact that all h_{∞} values have been obtained with 50-msec prepulses, which may be insufficient at low temperature. Figure 6C also gives evidence that 50-msec prepulses are too short at low temperature. At low temperature, the $h_{\infty}(E)$ curve obtained with 250-msec prepulses is flatter than the $h_{\infty}(E)$ curve obtained with 50-msec prepulses, which is shown for comparison, and does not reach the abscissa for depolarizing prepulses. Thus, the present data are in accordance with the assumption that E_h varies linearly with T.

Discussion

The main results of the present paper are: (i) τ_{on} of the gating current and τ_m of the Na permeability are highly temperature dependent. The maximum amount of displaced charge is only weakly temperature dependent. (ii) The time constant of charge immobilization, τ_i , is highly temperature dependent. The Q_{10} value for τ_i^{-1} is 2.71. τ_i and the time constant of inactivation, τ_h , are similar within the investigated temperature range. (iii) Temperature changes cause shifts of stationary curves $Q_{on}(E)$, $P_{Na}(E)$, and $h_{\infty}(E)$.

COMPARISON WITH PREVIOUS RESULTS

The Q_{10} value of 2.54 for the time constant of the on-response of the asymmetry current reported in this paper is between the values reported by Bezanilla and Taylor (1978) and Kimura and Meves (1979) for squid axon. Bezanilla and Taylor (1978) found a Q_{10} of 1.6 for early and 2.3 for late components of the on-response of the asymmetry current between 6 and 16°C. Kimura and Meves (1979) measured Q_{10} values between 6.4 and 2.0 in the temperature range 0–15°C. The Q_{10} given above is slightly larger than the values given in short reports of Schauf and Bullock (1979) for *Myxicola* giant axons and of Nonner et al. (1975) and Collins and Rojas (1982) for myelinated nerve. The Q_{10} values of these three papers range between 2.0 and 2.4.

The weak temperature dependence of the amount of displaced charge is in accordance with the results of Bezanilla and Taylor (1978) obtained from squid axon ($Q_{10} = 1.13$ between 5 and 26°C, charge measured between E = -100 mV and E = 50 mV). Kimura and Meves (1979), however, reported considerably higher Q_{10} values for Q_{on} in squid axon ($Q_{10} = 1.41$ between 0 and 15°C, measured with test pulses to E = 0 or E = 20 mV). The reasons for this

discrepancy may be: (i) Kimura and Meves worked in a lower temperature range and (ii) Kimura and Meves determined Q_{on} with less positive pulses than Bezanilla and Taylor and I. As pointed out by Kimura and Meves (1979), shifts of the $Q_{on}(E)$ curves to the left with increasing temperature could pretend a relatively high Q_{10} value for $Q_{on}(E)$ obtained with test pulses of small amplitude, although the "real" Q_{10} value for Q_{max} may be near one.

The present paper shows that the kinetics of charge immobilization is temperature dependent and that the time constants of charge immobilization and the time constants of fast inactivation are similar within a large temperature range. Two major arguments have been put forward in favor of the idea that charge immobilization and inactivation are more or less directly related: both have similar kinetics and potential dependence (Armstrong & Bezanilla, 1977; Nonner et al., 1978; Nonner, 1980; but see Meves & Vogel, 1977) and are similarly affected by neurotoxins (Nonner, 1979; but see Tanguy & Yeh, 1988). The similar temperature dependence of immobilization and inactivation demonstrated in the present paper is a third argument that immobilization and inactivation are strongly related, or even reflections of the same molecular process.

Shifts of $h_{\infty}(E)$ curves towards positive potentials with increasing temperature have been reported by Kimura and Meves (1979) for squid giant axon, by Chiu et al. (1979) for frog nodes and mammalian nodes and by Schwarz (1986) for mammalian nodes. Kimura and Meves found shifts of 7.3 mV (0 $vs. 15^{\circ}$ C), Chiu et al. of 5–19 mV (0 $vs. 25^{\circ}$ C), and Schwarz of 11.2 mV (0 $vs. 20^{\circ}$ C). The shift of 0.570 mV K^{-1} reported above is in good agreement with the previous data, especially considering that all reports except the paper from Chiu et al. (1979) deal with different preparations.

Shifts of $P_{\text{Na}}(E)$ curves towards positive potentials with increasing temperature have been reported by some authors but were not found by others. Kimura and Meves (1979) observed shifts of 7 mV (9.2 vs. 15.4°C), Schwarz (1986) of 5.5 mV (0 vs. 20°C), whereas Chiu et al. (1979) did not find any shift ($\leq \pm 10 \text{ mV}$) of $P_{\text{Na}}(E)$. The shift of 0.429 mV K^{-1} reported in the present paper is close to the mean of all. The discrepancy between these findings may be at least partly due to series resistance. The influence of R_s should be smaller in my experiments performed in Ri-TEA containing 5 nM TTX than in the experiments of Schwarz (1986) and Chiu et al. (1979) performed in full-Na solution without TTX. $Q_{on}(E)$ curves obtained from squid axon have been presented by Bezanilla and Taylor (1978) and by Kimura and Meves (1979). A shift of approximately 10 mV with increasing temperature (10 vs. 22° C) can

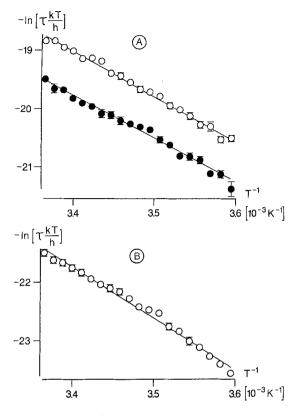


Fig. 7. Interpretation of the kinetic data with Eyring rate theory. $-\ln(\tau \cdot kT/h)$ was plotted *vs.* the reciprocal value of absolute temperature, *T*. The test pulse potential is the same (E = 4 mV) for all measurements. $(A) \tau = \tau_m (\bigcirc), \tau = \tau_{on} (\bullet)$. $(B) \tau = \tau_h$. The straight lines are drawn according to the results of linear regression yielding the following apparent values of activation enthalpy ΔH_a and activation entropy ΔS_a :

	ΔH_a (kJ mol ⁻¹)	$\Delta S_a \; (\mathrm{J} \; \mathrm{mol}^{-1} \; K^{-1})$	mean $Q_{10} \pm s_{\text{EM}}$
$ au_{\mathrm{on}}$	61.3	44.0	2.54 ± 0.07
τ_m	64.2	60.2	2.64 ± 0.06
$ au_h$	73.1	67.8	3.00 ± 0.09

The Q_{10} values shown in the table were calculated from $\ln(\tau)$ versus T plots. The Figure contains data from 394 Na current records (5 experiments) and 99 gating current records (7 experiments)

be seen in Fig. 1*B* of Bezanilla and Taylor (1978), but was not further discussed by the authors. Kimura and Meves could not decide whether the high Q_{10} for Q_{on} they observed was due to an increase of Q_{max} or due to a shift of the $Q_{on}(E)$ curve towards negative potentials. This uncertainty arises from the fact that—as can be seen in their Fig. 9—the saturation of charge at high positive potentials is not pronounced.

It is surprising that $Q_{on}(E)$ curves are shifted to the left, whereas $P_{Na}(E)$ curves are shifted to the

right when temperature is increased. A similar observation has been reported by Conti et al. (1982, 1984), who found that increasing pressure shifts $P_{\rm Na}(E)$ towards positive potentials, but does not shift $Q_{on}(E)$. The results of the present paper and the findings of Conti et al. (1982, 1984) can be explained with the hypothesis that transitions between the closed states of the Na channel contribute far more to the total charge movement than the final step preceding the opening of the Na channel. Indeed, there is a lot of evidence in favor of this hypothesis. The Na channel models of Armstrong and Gilly (1979) and Horn and Vandenberg (1984) assume this final step to be rate limiting; consequently, the corresponding component of gating current may have been lost in the baseline. Other models even postulate that this step is electrically silent (Edmonds, 1987). Therefore, the assumption that the final step makes negligible contribution to gating current seems to be justified and will be also made in the following model.

THERMODYNAMIC INTERPRETATION OF THE KINETIC DATA AND THE SHIFTS

The interpretation of the present results is complicated by the fact that there is no universal model of Na channel gating. It is controversal how many states exist, how they are linked by reaction steps, and which steps are slow and which are fast.

If one accepts that the time constant of gating current and the time constants of activation and inactivation of Na current each reflect mainly one reaction step, their temperature dependence may be explained in terms of Eyring rate theory. For a chemical reaction with a single energy barrier, rate constant r, standard activation enthalpy ΔH_a , and standard activation entropy ΔS_a are related as follows (*see* Tsien & Noble, 1969; Marshall, 1978):

$$-\ln(1/r \cdot kT/h) = -\Delta H_a/RT + \Delta S_a/R.$$
 (5)

k is the Boltzmann constant, h the Planck constant, R the gas constant, and T the absolute temperature. If the backward rate constants are much smaller than forward rate constants (this should be the case at E = 4 mV), one may substitute 1/r by the relaxation time constant τ . Figure 7 shows a plot of $-\ln(\tau \cdot kT/h)$ versus 1/T for τ_{on} , τ_{m} , and τ_{h} . All relations can be satisfactorily fitted with straight lines in the temperature range from 5 to 24°C. The values of ΔH_a and ΔS_a obtained from these plots (see legend of Fig. 7) are apparent values, because one does not exactly know how many reaction steps contribute to the observed relaxation time constant and how many energy barriers are crossed. Nevertheless, it is interesting to compare these apparent activation entropy values with the values of Tsien and Noble (1969), which were calculated from the Q₁₀ values of Frankenhaeuser and Moore (1963). The data of the present paper indicate that the apparent activation entropy values ΔS_a obtained from τ_{on} , τ_m , and τ_h are positive (*see* legend of Fig. 7). Tsien and Noble (1969) report a similar ΔS_a value for the inactivation process, but a completely different, i.e., negative, value for the activation process. The reason for this discrepancy is the low temperature dependence of τ_m in the data set of Frankenhaeuser and Moore (1963, Q₁₀ = 1.84 for α_m).

The temperature dependent shifts of Q_{on} , P_{Na} , and the fraction of inactivated channels $(1-h_{\infty})$ which are assumed to be steady-state parameters and are symbolized as x_{∞} , can be explained as follows: assuming two-state gating mechanism $A \leftrightarrow B$, the electrochemical equilibrium between these two states can be described by the function

$$x_{\alpha}(E) = 1/\{1 + \exp[(\Delta G - z \cdot F \cdot E)/RT]\}$$
(6)

where $x_{\infty}(E)$ is the fraction of "gates" in state B, $\Delta G = G_B - G_A$ (which are the standard free energy values of B and A in the absence of electric field) and z is the number of elementary charges associated with the transition $A \rightarrow B$ (see Marshall, 1978). With $\Delta G = \Delta H - T \cdot \Delta S$, where ΔH and ΔS denote standard enthalpy and entropy differences, respectively, the midpoint potential E_h of the curve (where $x_{\infty} = 0.5$) can be calculated as

$$E_h = -T \cdot \Delta S/zF + \Delta H/zF. \tag{7}$$

Temperature dependence of E_h is obvious if ΔS is different from 0. Finally, one may calculate $\Delta S =$ $-zF \cdot \Delta E_h/\Delta T$; this approach, however, is strictly valid only if ΔH and ΔS are independent of temperature. A variation of ΔH and ΔS with temperature, which has been reported for lysozyme (Pfeil & Privalov, 1976), would predict a change of dE_h/dT with temperature. No such change, however, was found experimentally: $Q_{on}(E)$, $P_{Na}(E)$, and $h_{\infty}(E)$ curves at intermediate temperature are approximately in the middle between the curves at high and low temperature. The shape of $h_{\infty}(E = -70 \text{ mV}, T)$ dT, as mentioned above. Thus, ΔH and ΔS of the Na channel protein seem to be, in first approximation, constant within the investigated temperature range.

If one assumes that the final step opening the Na channel makes only little contribution to the gating current (*see* above), an entropy profile of Na channel gating can now be deduced from the ob-

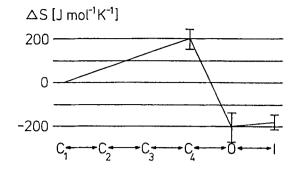


Fig. 8. Interpretation of the equilibrium data: A hypothetical entropy profile of Na channel. Entropy value ΔS of a certain state relative to C_1 was calculated from the mean shifts, the mean slope factors, and the mean temperatures in Table 2 according to Eq. (7). Handling the shifts of $Q_{on}(E)$ and $h_{x}(E)$ is straightforward, because the experimental data are fitted conveniently with Eq. (6). From the shift of the $Q_{on}(E)$ curve, the average ΔS for a transition $C_i \rightarrow C_{i+1}$ can be calculated as $\Delta S = -zF \Delta E_h/\Delta T = 66$ J mol⁻¹ K^{-1} . From the shift of the $h_{\infty}(E)$ curve, ΔS for the transition $C_1 \rightarrow I$ can be calculated as $-182 \text{ J mol}^{-1} K^{-1}$. Treatment of the shift of the $P_{Na}(E)$ curve is more difficult, because $P_{Na}(E)$ curves are not conveniently fitted with Eq. (6). The curve is asymmetric and contains information about the step $C_4 \rightarrow 0$ towards positive potentials and about the transitions $C_1 \rightarrow 0$ towards negative potentials (Almers, 1978). Shifts and steepness factors were obtained from fits in $\ln(P_{Na})$ versus E plots, which should weight the data in the negative potential range relatively high and therefore describe the $C_1 \rightarrow \rightarrow 0$ transitions; the corresponding ΔS is calculated as $-201 \text{ J mol}^{-1} K^{-1}$. The assumption of four closed states, i.e., four reaction steps before the open state is reached is based on the observation that k^* for $O_{cr}(E)$ is 3.6 times as large as for $P_{Na}(E)$, but is somewhat arbitrary. It is assumed that each of the three reaction steps between the closed states leads to the same change of entropy ΔS . Bypass reactions, which could, for example, lead from C_4 directly to I (Horn & Vandenberg, 1984), are not displayed. They should, however, have no consequences for the estimation of the entropy profile, because introducing a bypass reaction does not change the equilibrium if the system still obeys microscopic reversibility

served shift values. Such a hypothetical entropy *vs.* state diagram, based on a Markovian model with four closed states, is given in Fig. 8; details of the calculation procedure are described in the legend of the figure.

Different interpretations of the shifts induced by temperature changes are, however, possible.

i) Screening of surface charges on the outer side of the membrane may depend on temperature, because the equation of Grahame (1947) includes absolute temperature T as well as the dielectric constant of water, which is itself a function of T (Lax, 1967). The shifts of potential-dependent parameters calculated from Grahame's equation using the surface charge density of Vogel (1974) are, however, by a factor of about two smaller than the observed shifts. Furthermore, the screening hypothesis can hardly explain why increasing temperature shifts Q_{on} , on one hand, and P_{Na} and h_{∞} , on the other hand, into different directions.

ii) A phase transition of the membrane lipids could occur, which might change the equilibrium of reactions of the Na channel. This seems unlikely, since the shifts are observed within the whole investigated temperature range $5-25^{\circ}$ C, whereas transition temperatures are below 10°C in frog nerve (Schwarz, 1979).

iii) Decreasing temperature may select a population of channels with steady-state properties different from the average. Matteson and Armstrong (1982) postulated "sleepy" channels at low temperature (1°C) in squid axon. They found, however, similar $P_{Na}(E)$ characteristics for "sleepy" channels and normal channels. As the amount of gating charge is constant within the investigated temperature range (Fig. 1D), "sleepy" channels—if existing in the node of Ranvier at about 6°C—may have a "nonsleepy" gating machinery.

Consequences for Models of Na Channel Gating

Figure 8 shows that considerable entropy changes may occur when a Na channel passes from closed through open to inactivated states. Howarth (1975) observed measurable heat production during an action potential and explained this phenomenon with large entropy changes of the membrane, too. Baumann (1978) reported ΔS_a and ΔS values in the same order of magnitude for the conversion between metarhodopsin I and II ($\Delta S_a = 230 \text{ J mol}^{-1}$ K^{-1} and $\Delta S = 142$ J mol⁻¹ K^{-1}). Considering the lower molecular weight of metarhodopsin, the entropy changes seem to be even larger in this molecule. Entropy is a measure of the disorder of a system, e.g., the Na channel including its lipid and aqueous environment. Figure 8 therefore suggests that the intermediate closed states of the Na channel are states of low order, whereas the open and the inactivated state are states of high order. The protein structure may initially unfold and then collapse. It would be interesting to test whether recent hypotheses concerning the molecular mechanisms of Na channel gating are in accordance with the entropy profile given in the present paper.

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